

NMI3 JRA Sample Environment Meeting

Helmholtz-Zentrum Berlin (Wannsee)

High Pressure Task

22.11.2012





Activities, Deliverables & Milestones

Gas handling and pressure Cells for Inert Gases

- *D21.02* Report on current inert gas pressure cell technology (STFC) *Completed*
- *D21.06* 13-15 kbar hydraulic intensifier for cell testing assembly (STFC) *Completed*
- D21.07 Procurement of LLB cryogenic system for pressure measurements (LLB) Completed
- **D21.15** Procurement **10 kbar** automated gas handling system (STFC) Completed
- *D21.18* Manufacture a cell for up to 8 kbar (LLB) *Completed*
- *D21.26* Procure 10 kbar automated gas handling system (LLB) *Completed*
- **D21.29** Manufacture and test prototype cell up to 10 kbar (LLB) *Prototype has been manufactured; awaiting for test*





Activities, Deliverables & Milestones

Gas handling and pressure Cells for Hydrogen

- D21.08 Manufacture and test prototype cell for 4kbar and up to 700K (STFC) Completed
- **D21.12** Sourcing, assembly and commissioning of 8-10 kbar H₂ intensifier and gas handling system (STFC) *Work in progress* •••
- *M21.2.2.1* Cell material and seal design review *Completed*
- **D21.16** Report on material research: H₂ and neutron compatibility . *Completea*
- *M21.2.2.2* Design plan review for H_2 gas pressure cell for 8 kbar at 4-300K *Complete*
- *D21.24* Manufacture and test prototype cell for 6kbar and up to 300K (STFC) Completed
- D21.29 Manufacture and test prototype cell for 8 kbar 4 300K (STFC) Prototype has been manufactured; awaiting for test





D21.15 Procurement 10 kbar automated gas handling system



The automated inert gas handling system specification:

- Operating from 0 to 10,000bar
- Transducer accuracy of 0.3%
- Full scale and pressure changes in steps of 40 bar
- This system is able to be controlled by instrument computer

The system is released to ISIS User Program





D21.15 Sourcing, assembly and commissioning of 10 kbar H₂ Intensifier and gas handling system

Due to financial restraints the 10Kbar Hydrogen Intensifier is to be assembled and tested at ISIS by the Pressure & Furnace department. Components have been purchased and assembly work is in progress . *ISIS immensely benefitted from HZB experience*!



High pressure hydrogen stick



High pressure hydrogen "red" panel 9.5 kbar with Helium has been achieved





M21.2.2.1 Cell material and seal design review

10 kbar seal-test containment vessel



Typical Bridgman seal arrangement showing a three part seal assembly and the supporting backing nut.



Axial FEA results of the seal-test containment vessel.

Yanling Ma et. al. International Journal of High Pressure Research 32 (2012) 364





M21.2.2.1 Cell material and seal design review

JRA Bridgman seal test results



Seal Configuration	Room Temp Leak Tight at 2Kbar	Room Temp Leak Tight at 10Kbar	Liquid Nitrogen Leak Tight at 2Kbar	Liquid Nitrogen Leak Test at 10Kbar
Pb\Cu\Pb\Cu\Pb (Copper seals lead plated)	10 mins hold time	3hrs hold time ✓	10mins RT hold time then immersed in N_2 10mins hold time	Pressurised to 10Kbar in N_2 3hrs hold time
Al\Cu\Al\Cu\Al (Copper seals lead plated)	10 mins hold time ✓	Left over night hold time	10mins RT hold time then immersed in N ₂ 10mins hold time (slight leak observed) ✓	Pressurised to 10Kbar in N_2 (leak sealed) 3hrs hold time
Al\Cu\Al\Cu\Al (Copper seals <i>not</i> lead plated)	Seals struggled to seal but finally sealed at 2Kbar ✓	Seal remained leak tight until 7.6Kbar then failed and never resealed during attempt to obtain 10Kbar	Test not performed	Test not performed



Aluminium/copper seal combination (left); Lead/copper seal combination (right).





M21.2.2.1 Cell material and seal design review

D21.16 Report on material research: H_2 and neutron compatibility

Hydrogen compatible material tests Joint project with *Imperial College, the University of London*



ence & Technology

Hydrogen embritlement is a process where certain materials become brittle following exposure to hydrogen. High-strength steels, titanium alloys and aluminium alloys seem particularly vulnerable to this.



17-4PH tensile test sample



17-4PH tensile test sample



Fatigue results of various vessel materials





Hydrogen gas cells



D21.24 **6kbar** and **7kbar** Hydrogen gas cells for **LT – 300 K**



D21.29 Prototype cell for **8 kbar** of Hydrogen **LT – 300 K**



High magnetic field + *high pressure*



Free energy landscape of hydrogenated La(Fe,Si)13 K G Sandeman, Z G Gercsi, Imperial College London, UK O Gutfleisch, IFW-Dresden, Germany

A. Barcza et al., "Giant Magnetoelastic Coupling in a Metallic Helical Metamagnet" Phys. Rev. Lett. 104 247202 (2010) [HRPD + theory]



7.5T magnet



High magnetic field + high pressure



We seem to be observing the influence of the pressure-temperature-field-induced critical point (i.e. first order changing to second order transition in a single material, depending on the exact experimental conditions).





Screenshot 1 P=7kbar, 1st order transition at 20 K above "Tc" with increasing field (B=4.5 to 5.5T @235K)

By courtesy of Dr. Karl Sandeman, Imperial College, University of London Screenshot 2 P=0, 2nd order transition at 15K above Curie temperature with increasing field (B=0-5T@285K)







The University of Edinburgh School of Engineering CSEC Konstantin Kamenev's Group

"SPUTNIK" Cell:

- up to 6 Gpa (60 kbar)
- 0.05 300K



SPUTNIK - 3





JRA SE High Pressure Task technical publications:

- 1. Yanling Ma et. al. International Journal of High Pressure Research 32 (2012) 364
- 2. O. Kirichek et. al. Journal of Physics: Conference Series 340 (2012) 012008
- 3. R. Done et.al. High pressure gas vessels for neutron scattering experiments *ArXiv* (2010) 1007.3135
- 4. R. Done et.al. *NMI3 FP7 JRA Report* on current inert gas pressure cell technology (2010)

Experiments which directly benefited from JRA SE project:

- E. Gray, C.J. Webb In-situ diffraction techniques for studying hydrogen storage materials under high hydrogen pressure *International Journal of Hydrogen Energy* 37, (2012) 10182
- 2. M. T. F. Telling et. al. *Phys. Rev. B* 85 (2012) 184416
- 3. T. Lancaster et. al. *Phys. Rev. B* 84 (2011) 092405
- 4. M Enomoto et. al. *Polyhedron* **30**, (2011), 3137
- 5. F K Katsaros et al Journal of Physics: Conference Series 340 (2012) 012046







Experiments which benefited from technologies and equipment developed by JRA SE project:

Sihai Yang et. al. *Selectivity and Direct Visualization of Carbon Dioxide and Sulfur dioxide in a Decorated Porous Host Nature Chemistry* (2012) doi:10.1038/nchem.1457

Kazuya Kamazawa et. al. *In-Operando Neutron Diffraction Studies of Transition Metal Hydrogen Storage Materials Advanced Energy Materials* (2012) 10.1002/ aenm.201200390









Acknowledgements

We are grateful to our partners from ILL, HZB, LLB, PSI and FRM II for really fruitful collaboration in NMI3 FP7 JRA SE framework supported by the European Commission.

We also really appreciate great opportunity of knowledge and experience sharing with our colleagues from neutron facilities all over the world.

Thank you very much for your attention!

