



Task 3: Humidity Chamber

NMI3-Soft Matter JRA-WP20

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27.11.2013 - Garching

In this talk...

- Project overview
- Proposed changes since INCS in July
- New (and almost final) design and accompanying simulations

Humidity control techniques

Saturated salt solution



Figure 4. Saturated salt humidity chamber. Hauß, V1, HZB.

- ✓ precise and reliable (tables available)
- ✓ no calibration necessary
- ✗ discrete humidity steps
- ✗ slow equilibration times

Gas vapour flow

- ✓ continuous humidity range possible
- ✓ automated humidity change (with mass flow controllers)
- ✓ fast equilibration time
- ✗ upper limit of humidity ~95%
- ✗ temperature gradients in cell or tubing could cause condensation

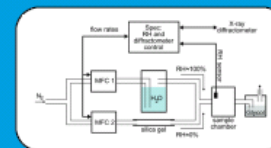


Figure 5. Humidity control setup Salditt, IRP.^a

Bulk water

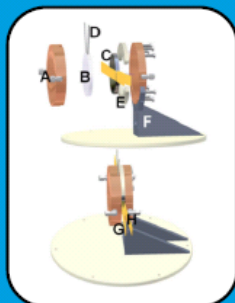


Figure 6. Reflectometry bulk water cell. Harroun, CINS.^g

- ✓ 100% relative humidity achievable
- ✓ quick deuterium contrast change in-situ
- ✗ sample loss to bulk solution (charged lipids)
- ✗ limited to reflectometry

Temperature controlled

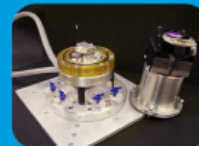


Figure 7. Temp. controlled cell Rheinstädter, McMaster.



Figure 8. Temp. controlled cell. Heinrich, NIST.

- ✓ high (>95% r.h.) possible
- ✓ quick deuterium contrast in-situ
- ✗ temperature gradients (from Peltier or external) lead to condensation
- ✗ difficult to calibrate heaters for desired r.h.

Humidity control techniques

	Humidity ceiling (at 25 C)	Humidity Stability	Automation for RH change	Equilibration time (after RH change)	Contrast variation (H ₂ O/D ₂ O)
Gas flow	~95%	~0.1% (or better?)	MFCs	minutes	Bubble through mix
Saturated salt	98% (discrete steps)	Weak temperature dependence	Syringe pumps (Sat. and distilled)	hours	Change of sample/ flow into reservoir
Temperature controlled	≤100%	~0.01% (or better?)	Peltier/water bath temp	hours	Flow liquid into reservoir
Fixed humid	Same as salt	NO regulation	X	X	Change of sample
Bulk water	Saturated	Stable	X	X	Flow into bulk volume

Design phase - working principle



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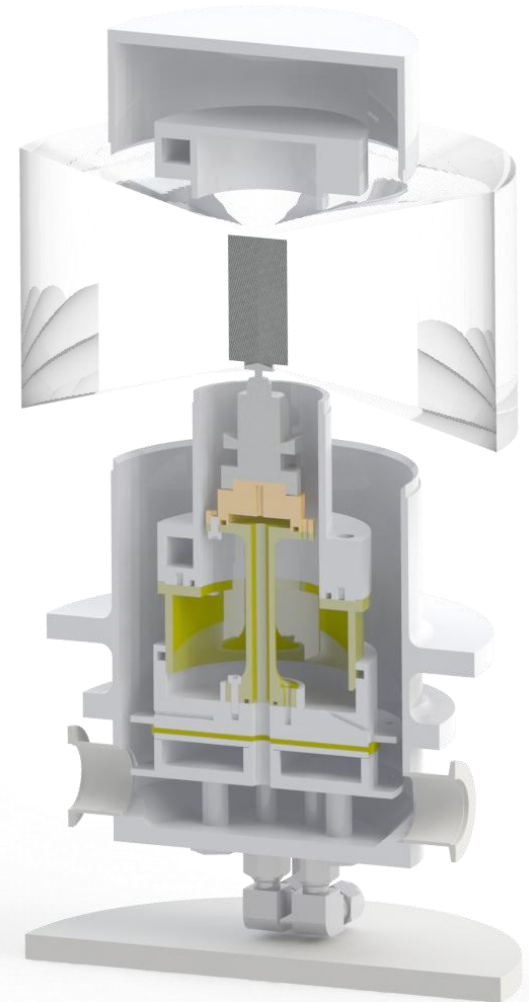
Bridgeman and Aldrich, 1964



Since July...

11.07.2013 - ICNS

Today

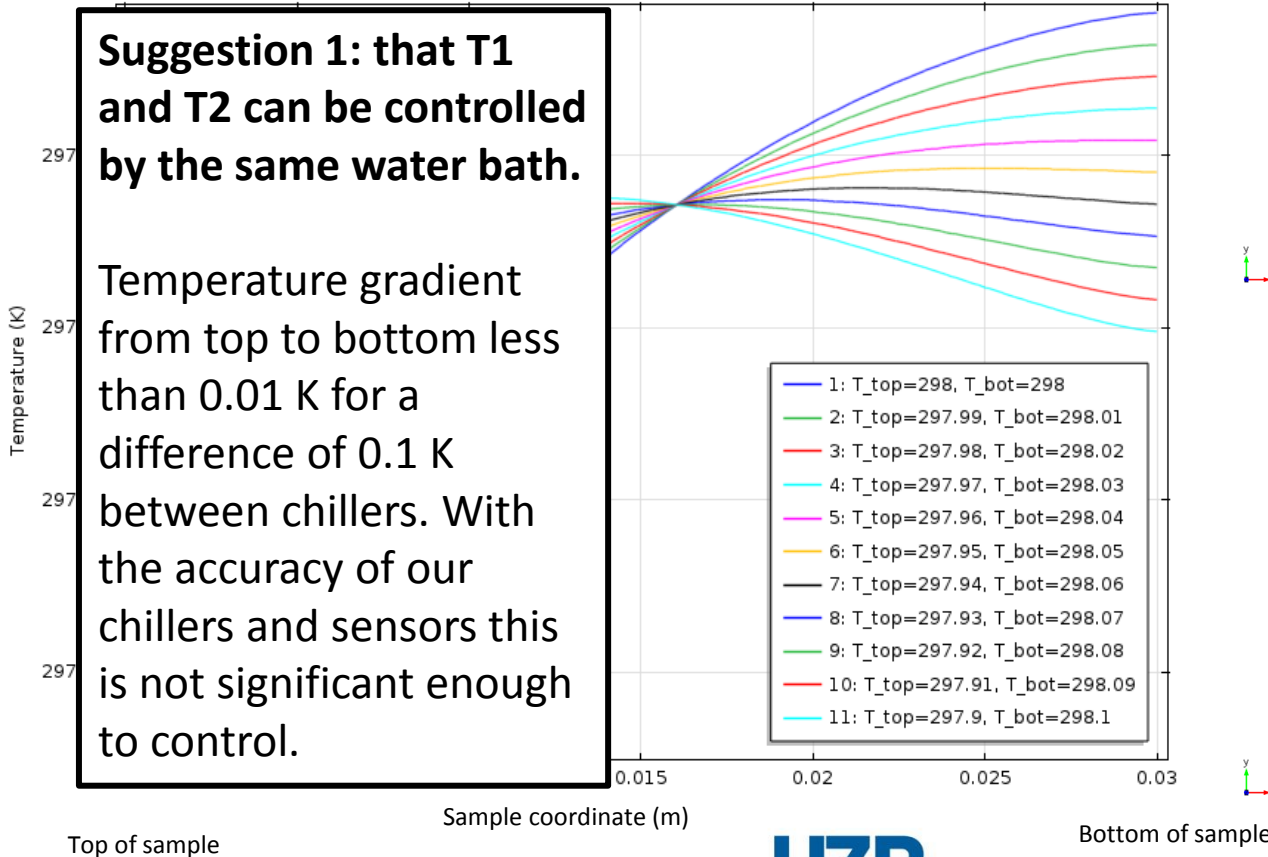


1. No T gradient in upper cell ($T_1=T_2$)

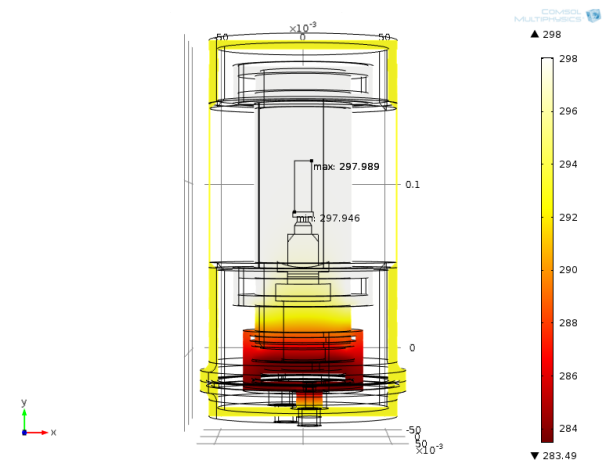
40% (r.h.) parametric study of bottom and top chiller temperature divergence

Sample Temperature

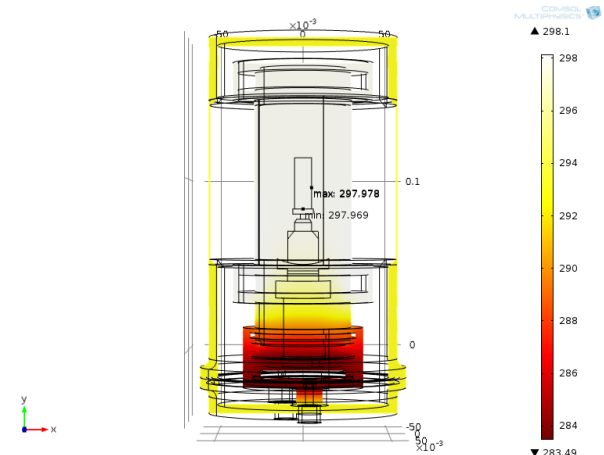
COMSOL MULTIPHYSICS



$T_1=T_2=298$ K

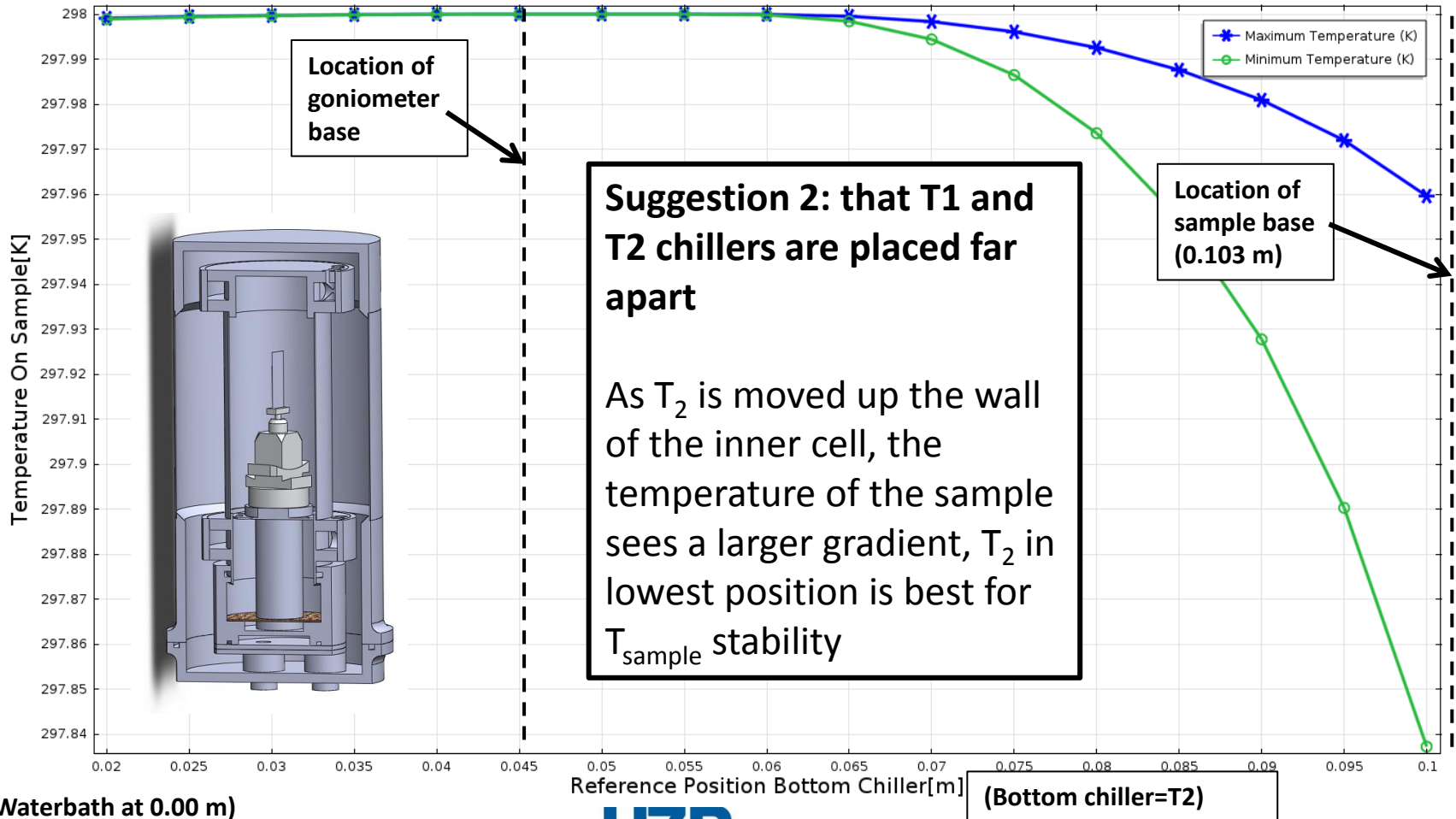


$T_1=298.1$ K $T_2=297.9$ K



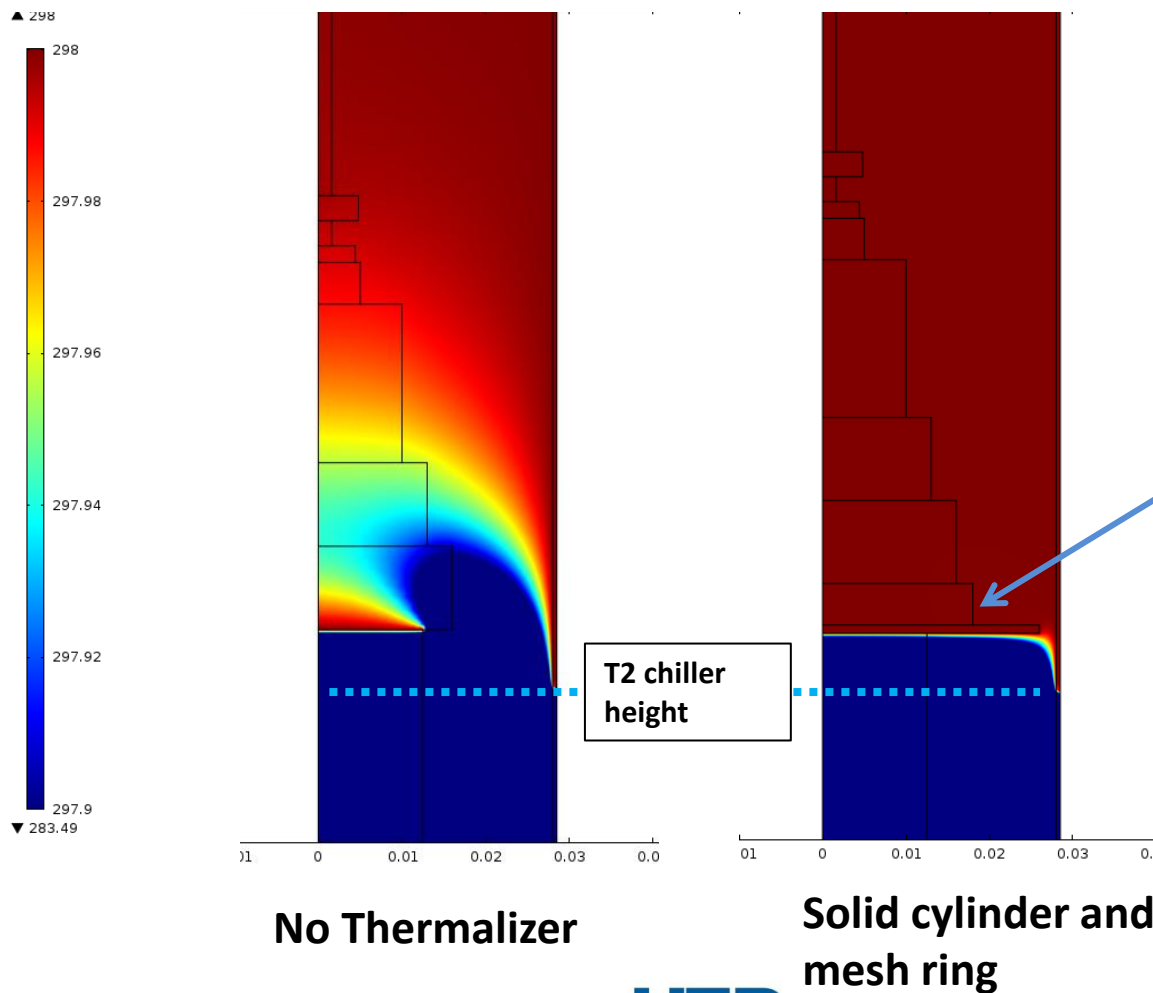
2. Stretched upper cell (max. distance between T1,T2)

40% rh parametric study of T2 chiller height, **plastic** goniometer base fixed at 0.048 m



4. 'Thermalizer' plate and mesh of high conductivity (placed above T2)

40% (r.h.) study of temperature gradients with and without plate and mesh

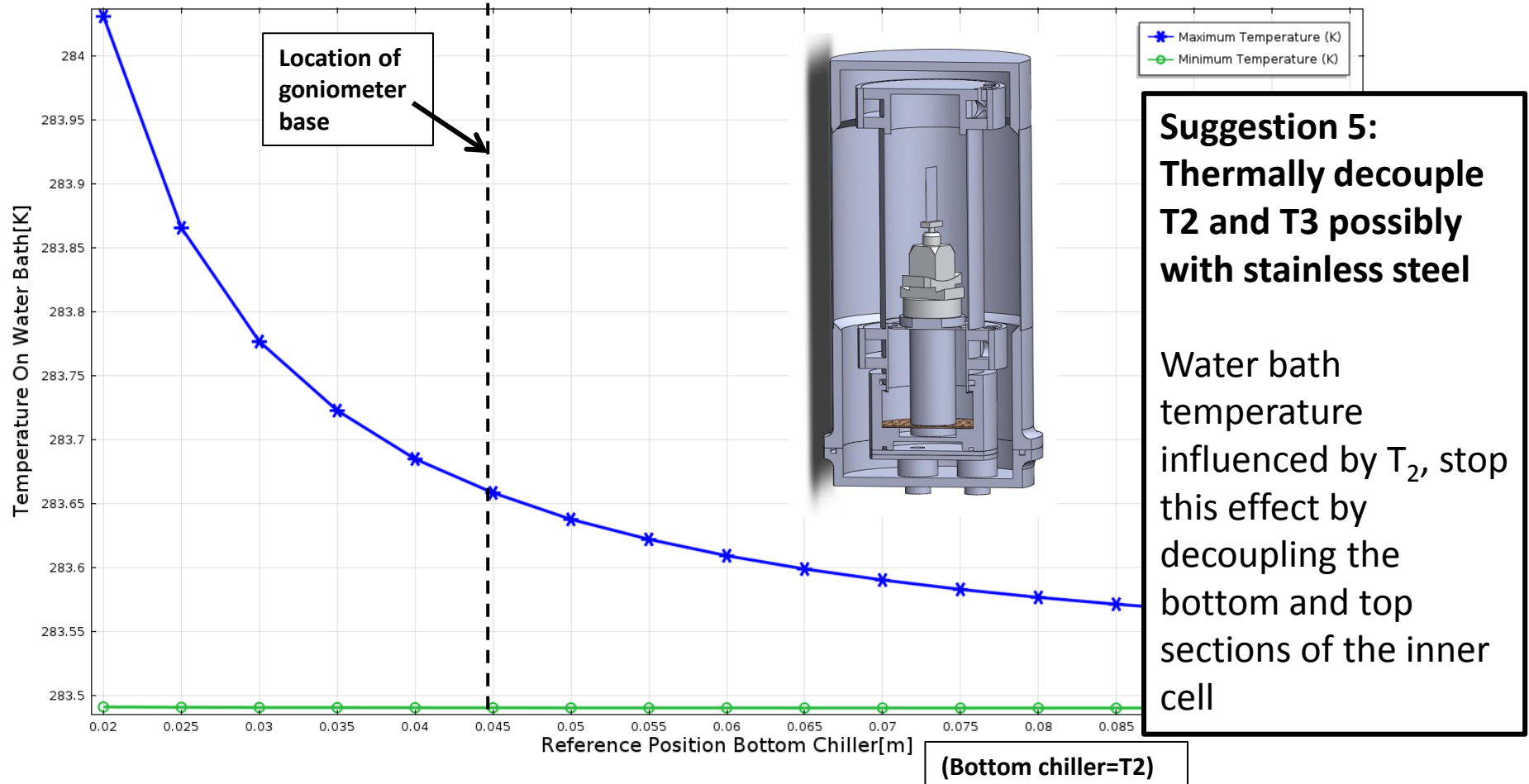


Suggestion 4: Add high conductivity material as thermalizer

Inhibit cold influence by spreading heat from heater using cylindrical plate and wide mesh ring

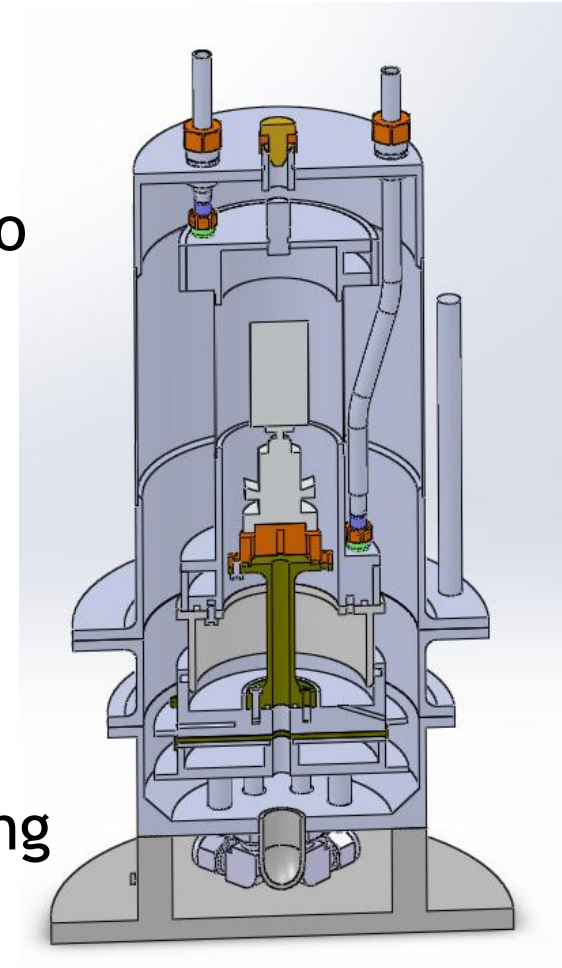
5. Redesign of bottom cell: completely decouple T_2 from T_3 (stainless steel?)

40% rh parametric study of T2 chiller height, goniometer base fixed at 0.048 m



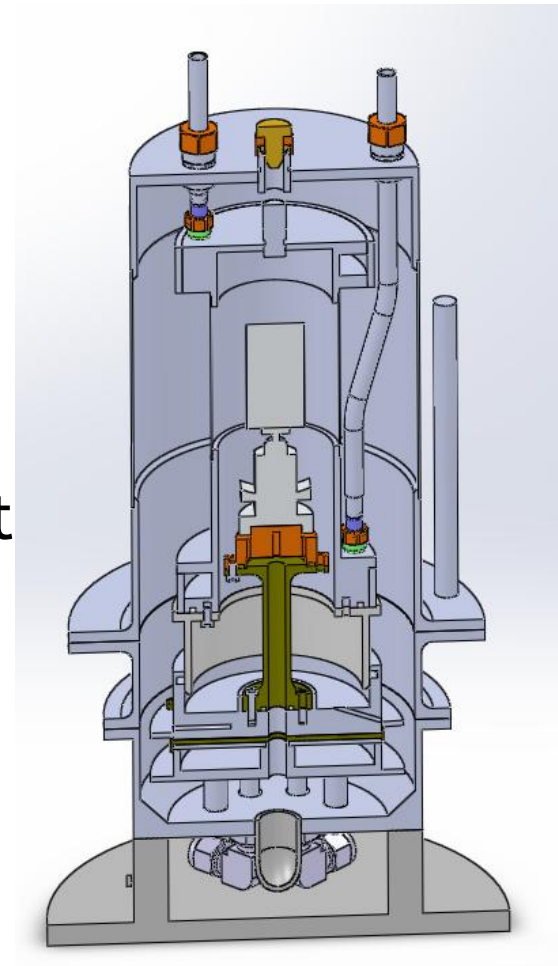
Inner can view

- Double walled (evacuated) Al
- Total size=400xØ110 (270 mm from base to sample) with
- Inner cell=220xØ50 (170 mm above cold chamber region)
- Wide angular scanning range possible ($\sim 300^\circ$), neutron windows with 15° opening

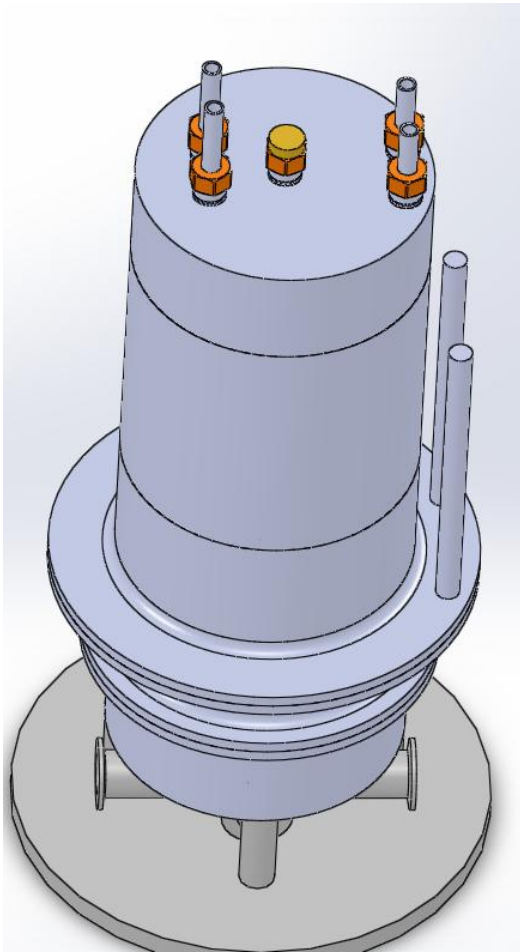


Inner can view

- 3 water chiller channels for precise temp.
- 2 resistive heating foils possible (below gonio and below reservoir)
- Sample cell thermally isolated (plastic post steel supports), T2 and T3 well thermally isolated
- Access for wiring above and below sample (T or RH sensors, Peltier element)

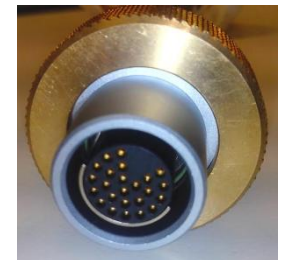
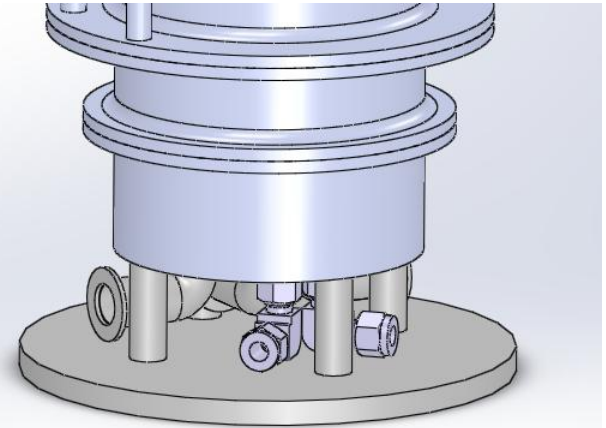


Outer can view



- Above: 4 Wilson seals for T1, T2 water chillers, option for sensors from above

- Below: 3 KF16 flanges and 2 Swagelok fittings for T3 water chiller, heating plates, and sensors



Jäger Connectors

Thanks



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Soft Matter JRA Observers



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